## Ba/Bs/EVS/C-2T



2024

(FYUGP)

(1st Semester)

## ENVIRONMENTAL SCIENCE

(Major)

Paper Code: EVS/C-2T

( Physics and Chemistry of Environment )

Full Marks: 75
Pass Marks: 40%

Time: 3 hours

The figures in the margin indicate full marks for the questions

## Answer any five questions

- 1. (a) Define light. Differentiate between monochromatic and polychromatic light, and mention two properties of light.

  1+2+2=5
  - (b) Define a blackbody. State Kirchhoff's law in relation to blackbody radiation.

1+4=5

(c) What is Rayleigh and Mie scattering?

Explain Beer-Lambert law. 1+1+3=5

L25/95

(Turn Over)

2.	(a)	Write notes on heat and work. $2\frac{1}{2}+2\frac{1}{2}=5$
	(b)	Define absolute temperature. Explain the relationship between temperature and kinetic energy. Write a note on thermal equilibrium. 1+2+2=5
	(c)	Elucidate on the working principle and efficiency of hydropower plant. 5
3.	(a)	Explain the Gaussian plume model. 5
	(b)	What are the sources of pollutants? Discuss pollutant dispersal. 2+3=5
	(c)	Write notes on mixing heights and turbulence. $2\frac{1}{2}+2\frac{1}{2}=5$
4.	(a)	Mention in detail the types of smog. 5
	(b)	Explain the chemistry of acid rain. 5
	(c)	Define aerosol with examples. Elaborate on the role of CFCs in ozone depletion.  2+3=5
5.	(a)	What is atomic structure? Specify Dalton's atomic theory. 1+4=5
	(b)	What are redox reactions? Explain its types. 1+4=5
	(c)	Draw a comparison between aliphatic and aromatic compounds. $2\frac{1}{2}+2\frac{1}{2}=5$
L25/	95	(Continued)

6.	(a)	Expound in detail the types of chemical bonds.
	(b)	Explain the working principle of a galvanic cell.
	(c)	Give a brief explanation on substitution and elimination reactions. $2\frac{1}{2}+2\frac{1}{2}=5$
7.	(a)	What is cohesion and adhesion? Explain the amphoteric nature of water. 2+3=5
	(b)	What is alkalinity of water? State the factors that affect water alkalinity. 1+4=5
	(c)	How is permanent hardness of water removed? Explain. 5
8.	(a)	What are complexes? Mention the significance of formation of complexes. 2+3=5
	(b)	Define colloids. What are the different types of colloids? 1+4=5
	(c)	Explain in brief the industrial and domestic sources of heavy metal contamination in water. $2\frac{1}{2}+2\frac{1}{2}=5$
9.	(a)	Explain soil composition. 5
	(b)	What is soil carbon? Mention the sources of soil carbon. 1+4=5
L25/	95	(Turn Over)

- (c) Mention the organic and inorganic components of soil.  $2\frac{1}{2}+2\frac{1}{2}=5$
- 10. (a) Explain ion exchange in soil. Write its importance. 3+2=5
  - (b) What are the factors affecting phosphorus availability in soil? Explain. 5
  - (c) Describe phenolics. What are the effects of phenolics on soil? 2+3=5

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